

Neutron crystallography to inform drug design targeting SARS-CoV-2 main protease

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SARS-CoV-2 life cycle

Proteolysis of pp1a and pp1ab by 3CL M^{pro} and PL^{pro} produces NSPs

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- Action by the two enzymes is vital for the viral replication cycle
- Inhibition of the proteases can stop the viral replication



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Main protease is the heart of SARS-CoV-2 replication

Stopping the heart of SARS-CoV-2

- 1) Polyproteins are cleaved into components of the replication machinery
- 2) Protease inhibitors bind active site, blocking substrate processing
- 3) Viral replication is prevented

Active site of SARS-CoV-2 Mpro





Substrate binding to M^{pro}

Targetable active site features

- Non-canonical catalytic Cys145-His41 dyad
- Room for ~6-7 peptide/inhibitor groups (P2'-P5)
- Characteristic oxyanion hole



M^{pro} active site: Where are the hydrogens?



Room-temp joint XN structure of M^{pro} in the native form @ 2.5Å resolution



Kneller et al. 2020 Nat. Commun. 18, 688-699 Kneller et al. 2020 J. Biol. Chem. 295, 17365-17373

Room-temp joint XN structure of Mpro-Telaprevir @ 2.4Å resolution



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Kneller et al. 2021 J. Med. Chem. 64, 4991-5000

M^{pro} active site is malleable adapting to ligand size Subsites S2 and S4 are cryptic



Tunability and malleability of M^{pro}

 Active site protonation states are tunable, but overall electrostatic charge is maintained at +1

 Active site conformation dynamically adapts to inhibitor properties

•Cryptic binding subsites and plasticity presents challenges for inhibitor design and in silico modeling







Structure-activity relationship (SAR) study for noncovalent inhibitors: Exploring structural, electrostatic and electronic determinants for binding to subsites S1 and S2



Supercomputer screen identifies noncovalent inhibitor



1 room temperature X-ray structure

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Clyde et al. 2022 J. Chem. Inf. Model. 62, 116-128

Room-temp joint XN structure of Mpro-1 complex @ 2.5Å resolution





Direct characterization of protein-ligand hydrogen bond network

Neutrons enable observation of new catalytic water molecule orientation

First neutron structure of M^{pro}-Non-covalent inhibitor complex



Neutron structure-led VR-assisted SAR



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SAR study guided by the neutron structure M^{pro}-1 complex (HL-3 series)



- > P1 must have H bond acceptor capability and correct geometry to make H bond with protonated His163.
- Substituents on the aromatic P2 group should have both moderate steric size and electronegativity.
- > P2 group with one substituent on the aromatic ring is disadvantageous.
- Highly electronegative substituents, such as -F or -CF₃ are disadvantageous, as are less electronegative but sterically larger ones.
- ➤ A third substituent such as -Cl at P2 group is favorable.

In vitro assessment of chemical modifications

Compound	Inhibition K _i (μM)	Affinity K _d (μM)	∆H (kcal/mol)	∆S (cal/mol⋅K)	∆G (kcal/mol)
MCULE-5948770040	2.9	1.30	-8.32	-0.7	-8.11
HL-3-68	0.89	0.69	-7.75	2.4	-8.5
Mcule-CSR-494190-S1	1.4	1.32	-9.1	-3.16	-8.15

HL-3-68 Mcule-CSR-494190-S1 Cl С Cl Ο 0 Ο -CH₃ -Cl C HN HN O= Cl O =C HN HN Ο

MCULE-5948770040



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M^{pro} is structurally and electronically malleable

- •One atom difference can significantly alter inhibitor binding potency
- •VR allowed true 3D structural analysis and inhibitor building
- •VR allowed inhibitor structures to be tailored to the binding site

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Design of covalent inhibitors based on hepatitis C virus protease inhibitors



Hepatitis C virus protease inhibitors bind and inhibit M^{pro}



Boceprevir





Narlaprevir









Design of covalent hybrid inhibitors of M^{pro}

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Unique binding of BBH-1 to M^{pro}: A neutron structure perspective

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Covalent inhibitors with nitrile warhead

Isothermal titration calorimetry

<u>Compound</u>	K _d ,	Stoichiometry,	ΔH ,	ΔS,	ΔG,
	μ M	N	kcal m ol ⁻¹	cal mol ⁻¹ K ⁻¹	kcal mol ⁻¹
BBH-2	0.030 ± 0.007	1.000 ± 0.005	-8.74 ± 0.08	5.40	-10.4
NBH-2	0.026 ± 0.016	0.990 ± 0.009	-8.76 ± 0.17	5.63	-10.5
PF-07321332	0.007 ± 0.003	0.990 ± 0.003	-10.75 ± 0.70	1.57	-11.2
GC-376	0.15 ± 0.03ª	0.99 ± 0.01	-6.7 ± 0.1	9.1	-9.4

Antiviral data

Hybrid inhibitors – fruitful path to clinical drugs

- Protonation states adapt to a specific inhibitor
- •Active site geometry adapts to inhibitor steric and electronic properties
- •Hybrid inhibitors are conceptually superior to previous designs

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